

## Past papers

Subject : Organic 233
Exam: Mid term
Done by: Abdallah Alkalaldeh

1-Assume that ( $2 \mathrm{~S}, 3 \mathrm{~S}$ ) -2,3-dichlorobutane has a specific rotation of $+14^{\circ}$. What would be the specific rotation of the ( $2 R, 3 R$ ) $-2,3$ - dichlorobutane:
A) $+14^{\circ}$
B) $0^{\circ}$
C) +7
D) $-14^{\circ}$
E) $+28^{\circ}$

2-Which of the following alkenes has Z - configuration ?
(Atomic number $\mathrm{H}=1 ; \mathrm{C}=6 ; \mathrm{N}=7,0=8, \mathrm{~F}=9$ )
A)IV
B) $V$
C)II
D)III

E)I

3-Which of the following compounds is chiral?
A)I
B)II
C)III
D)IV
E)V


I


IV


II


III

4-Which of the following groups have the highest priority according to priority rules? (atomic number H $=1 ; C=6 ; N=7 ; O=8)$ :
A) $-\mathrm{C} \equiv \mathrm{CH}$
B) $-\mathrm{CH}=\mathrm{CH}_{2}$
C) $-\mathrm{CH}_{2} \mathrm{NH}_{2}$
D) $-\mathrm{C} \equiv \mathrm{N}$
E) $-\mathrm{CH}=\mathrm{C}\left(\mathrm{CH}_{3}\right)_{2}$

5-If the observed optical rotation of an unknown sample is zero.
which of the following conclusions(s) is (are) true:
I) The sample is a racemic mixture
II) The sample is a meso compound
III) The sample is a pure enantiomer
A) I only
B) I and II only
C) II and III only
D) I and III only
E) Il only

6-The observed rotation for 100 ml . of an aqueous solution containing 6.0 g of sucrose, placed in a 2 -decimeter sample tube, is +1.2 at $25^{\circ} \mathrm{C}$. What is the specific rotation of Sucrose?
A) +50
B) +30
C) +20
D) +40
E) +10

7-The product (s) obtained from the following reaction is:
A)pair of diastereomers
B)racemic mixture
C) meso compound

D) pair of enantiomers
E)achiral

8 -What is the relationship between these two compounds?
A)diastereomers
B)enantiomers
C)identical
D)constitutional isomers



## E)conformations

9 -what are the possible number stereoisomers of ?
A) 5
B) 4
C) 2
D) 3



$\mathrm{OH} \mathrm{OH}^{*}$ [(1)
E) 1

10-What is the major product of the following reaction?

A) I
B) II
C) III
D) IV
E) V


11- What is the major product of the following reaction?

A) I
B) II
C) III
D) IV
E) V

12- Which of the following is (are) correct resonance structrure(s) for the intermediate formed in the nitration of bromobenzene :
A) I, II, IV AND V
B) III ONLY
C) I, II AND III
D) II ONLY
E) I ONLY


13-The intermediate in the nitration of nitrobenzene is:
A) I
B) II
C) III
D) IV
E) V

I

II

III

IV
V

14-which of the following names is correct :
A) p-aminochlorobenzene
B) m-nirtroaniline
C) p-bromodroxybenzene
D)chlorobenzoic acid-5
E) dichlorolbenzene-1,6

15- Which of the following statements about benzene is FALSE :
A) The bond angles are all 1200 and all carbon carbon bonds have the same length
B) The typical mechanism by which reactions occur is by addition
C) The molecule is planar and each carbon is at a corner of regular .hexagon
D) Each carbon in the benzene ring is sp2 hybridized
E) There are two resonance structures of equivalent energy

16- In an electrophilic aromatic substitution reaction, which group is both ortho, para ?directing, and ring deactivating:
A) anisole
B) chlorobenzene
C) benzoic acid
D) nitrobenzene
E) aniline

17- What is the major product of the following reaction?
A) I
B) II
C) III
D) IV
E) V


18- Which of the following compounds would be most reactive towards ring nitraration :
A) chlobenzene
B) benzene
C) nitrobenzene
D) phenol
E) benzoic acid

19- What is the major product of the following reaction?
A) I
B) II
C) III
D) IV
E) $V$


20- What is the major product of the following reaction?
A) I \& V
B) II \& III
C) I \& II
D) III \& V
E) IV

21- Which statement is true for SN2 reactions:

A) The rate of reaction is dependent only on substrate
B) The fastest reaction will occur with a tertiary halide
C) The mechanism is a two-step process
D) Substitution occurs with inversion of configuration
E) The rate of the reaction is dependent on the stability of a carbocation

22- Which reagent would you choose for the following reaction :

## $\mathrm{CH}, \mathrm{CH}, \mathrm{CH}, \mathrm{Br} \rightarrow \mathrm{CH}, \mathrm{CH}=\mathrm{CH}$,

A) -OH
B) $-\mathrm{Ch}_{3}(\mathrm{CO})_{3}$
C) -SH
D) $-\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{O}$
E) $-\mathrm{CH}_{3} \mathrm{O}$

23- Which of the following is an incorrect representation of relative nucleophile strength:
A) ${ }^{-} \mathrm{H}_{3} \mathrm{C}^{-}>\mathrm{HO}$
B) $\mathrm{HO}^{-}>\mathrm{HS}$
C) $\mathrm{CH}_{3} \mathrm{O}^{-}>\mathrm{CH}_{3} \mathrm{OH}$
D) ${ }^{-}>\mathrm{Br}$
E) $\mathrm{H}_{2} \mathrm{~N}^{-}>\mathrm{F}$

24- What is the leaving group in the following Reaction:
A) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{Br}$
B) $\mathrm{CH}_{3}-\mathrm{O}-\mathrm{CH}_{2} \mathrm{CH}_{3}$
C) Br
D) $\mathrm{Na}^{+}$
E) $\mathrm{CH}_{3} \mathrm{O}-\mathrm{NA}^{+}$
> $\rightarrow \mathrm{CH}_{3} \mathrm{O}^{-} \mathrm{Na}^{+}+\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{Br}$
> $+\mathrm{CH}_{3}-\mathrm{O}^{-} \mathrm{CH}_{2} \mathrm{CH}_{3}+\mathrm{Br}^{-}+\mathrm{Na}$

25- Name the following compound
A) carboxy-5-chloro-2-hexanone-6
B) chloro-6-oxoheptanoic acid-3
C) chloro-5-oxohexanoic acid-2
D) carboxy-3-chloro-5-hexanone-1
E) chloro-2-oxohexanoic acid-5


26-Which of the following is protic solvent
A) acetonitrile, $\mathrm{CH}_{3} \mathrm{C} \equiv \mathrm{N}$
B) dimethyl sulfoxide , $\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~S}=\mathrm{O}$
C) dimethylformamide, $\left(\mathrm{CH}_{3}\right)_{2} \mathrm{NCHO}$
D) acetone, $\left(\mathrm{CH}_{3}\right)_{2} \mathrm{C}=\mathrm{O}$
E) Methanol, $\mathrm{CH}_{3} \mathrm{OH}$

27 -Which would be the best way to carry out the following synthesis
A) $\mathrm{Br}_{2}(2) \mathrm{Mg}$, ether (3) $\mathrm{D}_{2} \mathrm{O}(1)$
B) $\mathrm{HBr}(2) \mathrm{Mg}$, ether (3) $\mathrm{D}_{2} \mathrm{O}$ (1)
C) $\mathrm{H}_{2} \mathrm{O} / \mathrm{H}^{*}(2) \mathrm{Mg}$, ether (3) $\mathrm{D}_{2} \mathrm{O}$ (1)
D) $\mathrm{D}_{2} \mathrm{O}$ (2) Mg , ether (1)
E) $\mathrm{D}_{2} \mathrm{O}$ (2) $\mathrm{Br}_{2}$, AlBr 3 (1)

28-What is the IUPAC name for the following molecule
A) ethyl isobutyl ether
B) ethoxy-3-methylpropane-3
C) ethoxybutane-3
D) butyl ethyl ether
E) ethoxybutane-2

$29 \rightarrow{ }^{*} \mathrm{CH} 3 \mathrm{CH} 2 \mathrm{MgBr}+\mathrm{CO} 2$ then $\mathrm{H}_{3} \mathrm{O}$
A) CH 3 COOH
B) CH 3 CH 2 OH
C) CH 3 CH 2 CH 3
D) CH 3 CH 2 COOH
E) CH 3 CH 2 CH 2 OH

30-The correct name for this molecule:
A) penten 2-methyl-2-ol-4
B) methyl-1-penten-2-ol-4
C) hydroxy-4-methyl-1-pentene-4

D) methyl-4-penten-2-ol-2
E) methyl-1-penten-4-ol-4

31-Which of the following is the weakest acid
A) ClCH 2 CH 2 COOH
B) CHCl 2 CH 2 COOH
C) CH 3 CCl 2 COOH
D) CH 3 CH 3 COOH
E) CH 3 CHClCOOH

32-which of the following alcohols would react most rapidly under SN1 conditions
A) CH 3$) 2 \mathrm{CHCH}, \mathrm{OH})$
B) CH 3 CH 2 OH
C) CH 3 CH 2 CH 2 OH
D) $\mathrm{CH} 3 \mathrm{CH} 2 \mathrm{CH}(\mathrm{CH} 3) \mathrm{OH}$
E) CH 3$) 3 \mathrm{COH}$ )

33 -The rate-determining step in the following reaction is

```
CH3)3
```

A) Ionization of alcohol to give carbocation
B) Displacement of water from the protonated alcohol by bromide ion
C) Protonation of alcohol
D) Capture of a carbocation by bromide ion
E) Loss of water from the protonated alcohol to give a carbocation

34-

A) 1
B) II
C) III
D) IV
E) V

35-The product(s) of the following reaction is (are) :
A) II \& III
B) I \& III
C) $I I$ ONLY
D) III ONLY
E) IONLY
F)

36-The product(s) of the following reaction is (are):
A) 1
B) 11
C) III
D) IV
E) V


37-
Which of the following reaction would give the following ketone ?

I) $\triangle \mathrm{C} \equiv \mathrm{CH} \xrightarrow[\mathrm{H}^{2+}]{\mathrm{H}^{+}, \mathrm{H}_{2} \mathrm{O}}$ II) $\longrightarrow \underset{\mathrm{H}}{\mathrm{C}}=\mathrm{CH}_{2} \xrightarrow{\mathrm{H}_{2} \mathrm{O} / \mathrm{H}^{+}}$
III) $\square \mathrm{CH}_{2}-\mathrm{CH}_{2} \mathrm{OH} \xrightarrow{\mathrm{CrO}_{3} / \mathrm{H}^{+}}$IV $\square \underset{\mathrm{H}}{\mathrm{C}}=\mathrm{CH}_{2} \frac{\text { 1) } \mathrm{O}_{3}}{\text { 2) } \mathrm{Zn} / \mathrm{H}^{+}}$
$v>-\mathrm{CH}_{2}-\mathrm{CH}_{2} \mathrm{OH} \xrightarrow{\mathrm{PCC}}$
A) 1
B) II
C) III
D) IV
E) $V$

38-Which of the following is hemiacetal :
A) 1
B) 11
C) III
D) IV
E) V

39-


A) 1
B) 11
C) III
D) IV
E) V

40-

A) 1
B) 11
C) III
D) IV
E) V

41-


42-

A) 1
B) 11
C) III
D) IV
E) V

Done

