

Subject : Organic 233 Exam: Mid term Done by: Abdallah Alkalaldeh

1-Assume that (2S , 3S) -2,3 - dichlorobutane has a specific rotation of + 14 $^{\circ}$. What would be the specific rotation of the (2R, 3R) -2,3 – dichlorobutane:

A) + 14 ° B)0 ° C)+ 7 D) -14 ° E) + 28 °

2-Which of the following alkenes has Z - configuration ? (Atomic number H = 1; C = 6; N = 7,0 = 8, F = 9)

A)IV B)V C)II D)III E)I I II III $H_2N \longrightarrow C = C \xrightarrow{CH_3} H_3C \xrightarrow{NH_2} OH$ IV V V

3-Which of the following compounds is chiral?

A)I B)II C)III D)IV E)V

4-Which of the following groups have the highest priority according to priority rules? (atomic number H = 1; C = 6; N = 7; O = 8) :

A) $-C \equiv CH$ B) $-CH = CH_2$ C) $-CH_2NH_2$ D) $-C \equiv N$ E) $-CH = C(CH_3)_2$

5-If the observed optical rotation of an unknown sample is zero. which of the following conclusions(s) is (are) true:

- I) The sample is a racemic mixture
- II) The sample is a meso compound
- III) The sample is a pure enantiomer

A) I only

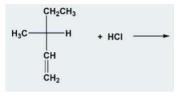
- B) I and II only
- C) II and III only
- D) I and III only
- E) II only

6-The observed rotation for 100 ml. of an aqueous solution containing 6.0g of sucrose, placed in a 2-decimeter sample tube, is +1.2 at 25°C. What is the specific rotation of Sucrose?

A)+ 50 B)+ 30 C)+ 20 D)+40 E)+10

7-The product (s) obtained from the following reaction is:

A)pair of diastereomers B)racemic mixture C)meso compound D)pair of enantiomers E)achiral



OH

Br

н

CH3

CH₃

8-What is the relationship between these two compounds?

A)diastereomers B)enantiomers C)identical D)constitutional isomers E)conformations

9-what are the possible number stereoisomers of ?

A)5	
B)4	
C)2	
D)3	
E)1	

FCH₂—CH₂—CH—CH—CH₂—CH₂F | | OH OH *

and

CH₃

Ĥ

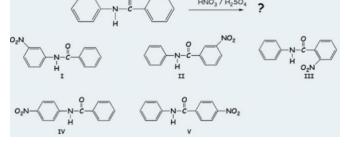
OH

CH₃

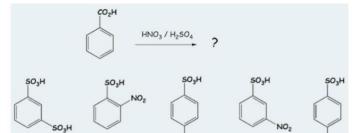
Br

10-What is the major product of the following reaction?

- A) I B) II
- C) III
- D) IV
- E) V



11- What is the major product of the following reaction?

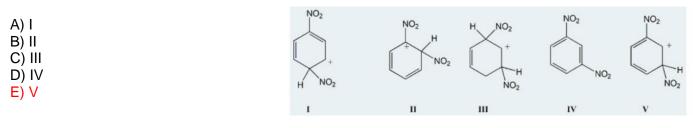


A) I B) II C) III D) IV E) V

12- Which of the following is (are) correct resonance structrure(s) for the intermediate formed in the nitration of bromobenzene :

A) I, II, IV AND V B) III ONLY C) I, II AND III D) II ONLY E) I ONLY	Br H NO ₂	H NO ₂	+ H NO ₂	H Br NO ₂	H Br NO ₂
	I	п	III	IV	v

13-The intermediate in the nitration of nitrobenzene is :



14-which of the following names is correct :

A) p-aminochlorobenzene

B) m-nirtroaniline

C) p-bromodroxybenzene

D)chlorobenzoic acid-5

E) dichlorolbenzene-1,6

15- Which of the following statements about benzene is FALSE :

A) The bond angles are all 1200 and all carbon carbon bonds have the same length

B) The typical mechanism by which reactions occur is by addition

C) The molecule is planar and each carbon is at a corner of regular .hexagon

D) Each carbon in the benzene ring is sp2 hybridized

E) There are two resonance structures of equivalent energy

16- In an electrophilic aromatic substitution reaction, which group is both ortho, para ?directing, and ring deactivating:

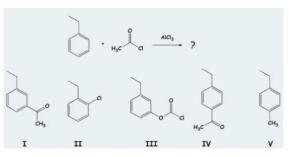
A) anisole

B) chlorobenzene

- C) benzoic acid
- D) nitrobenzene
- E) aniline

17- What is the major product of the following reaction?

- A) I
- B) II
- C) III
- D) IV
- E) V

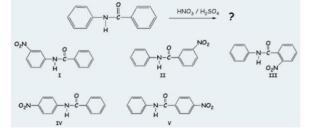


18- Which of the following compounds would be most reactive towards ring nitraration :

A) chlobenzene
B) benzene
C) nitrobenzene
D) phenol
E) benzoic acid

19- What is the major product of the following reaction?

- A) I
- B) II
- C) III
- D) IV
- E) V



20- What is the major product of the following reaction? A) I & V

- B) || & ||| C) | & || D) ||| & V
- E) IV

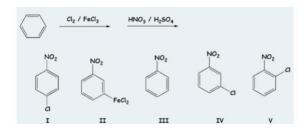
21- Which statement is true for SN2 reactions:

A) The rate of reaction is dependent only on substrate

- B) The fastest reaction will occur with a tertiary halide
- C) The mechanism is a two-step process
- D) Substitution occurs with inversion of configuration

E) The rate of the reaction is dependent on the stability of a carbocation

22- Which reagent would you choose for the following reaction :



$\textbf{CH,CH,CH,Br} \rightarrow \textbf{CH,CH=CH},$

A) -OH

B) -Ch₃(CO)₃

C) -SH

D) -CH₃CH₂O

E) -CH₃O

23- Which of the following is an incorrect representation of relative nucleophile strength:

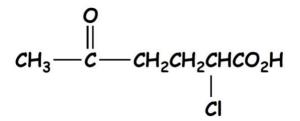
- A) $^{-}H_{3}C^{-} > HO$
- B) $^{-}HO^{-} > HS$ C) CH₃0⁻ > CH₃OH
- D) $|^{-1} > Br$
- E) $H_2N > F$

24- What is the leaving group in the following Reaction: A)CH₃CH₂Br B) CH₃-O-CH₂CH₃ C) Br $^{-}$ D) Na $^{+}$ E) CH₃O-NA $^{+}$

 \rightarrow CH₃O⁻Na⁺ + CH₃CH₂Br +CH₃-O-CH₂CH₃ + Br⁻ + Na

25- Name the following compound

A) carboxy-5-chloro-2-hexanone-6
B) chloro-6-oxoheptanoic acid-3
C) chloro-5-oxohexanoic acid-2
D) carboxy-3-chloro-5-hexanone-1
E) chloro-2-oxohexanoic acid-5



26-Which of the following is protic solvent

A) acetonitrile, $CH_3C\equiv N$ B) dimethyl sulfoxide , $(CH_3)_2S=O$ C) dimethylformamide, $(CH_3)_2NCHO$ D) acetone, $(CH_3)_2C=O$ E) Methanol, CH_3OH

27 -Which would be the best way to carry out the following synthesis

A) Br₂ (2) Mg, ether (3) D₂O (1)

$$\begin{array}{c} CH_{3}-CH=CH_{2} \rightarrow CH_{3}-CH-C \\ | \\ D \end{array}$$

- B) HBr (2) Mg, ether (3) D₂O (1)
- C) H_2O / H^* (2) Mg, ether (3) D_2O (1)
- D) $D_2O(2)$ Mg, ether (1)
- E) $D_2O(2)$ Br₂, AlBr3 (1)

28-What is the IUPAC name for the following molecule

- A) ethyl isobutyl ether
- B) ethoxy-3-methylpropane-3
- C) ethoxybutane-3
- D) butyl ethyl ether
- E) ethoxybutane-2

29- \rightarrow *CH3CH2MgBr + CO2 then H₃O

- A) CH3COOH
- B) CH3CH2OH
- C) CH3CH2CH3
- D) CH3CH2COOH
- E) CH3CH2CH2OH

30-The correct name for this molecule:

- A) penten 2-methyl-2-ol-4
- B) methyl-1-penten-2-ol-4
- C) hydroxy-4-methyl-1- pentene-4
- D) methyl-4-penten-2-ol-2
- E) methyl-1-penten-4-ol-4

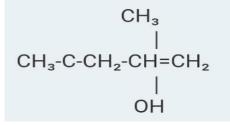
31-Which of the following is the weakest acid

- A) CICH2CH2COOH
- B) CHCI2CH2COOH
- C) CH3CCI2COOH
- D) CH3CH3COOH
- E) CH3CHCICOOH

32-which of the following alcohols would react most rapidly under SN1 conditions

- A) CH3)2CHCH,OH)
- B) CH3CH2OH
- C) CH3CH2CH2OH
- D) CH3CH2CH(CH3)OH
- E) CH3)3COH)

CH₃CH₂CH-O-CH₂CH₃ | CH₃

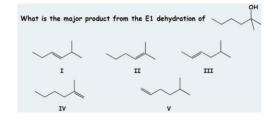


33-The rate-determining step in the following reaction is

 $CH_3)_3COH + HBr \rightarrow (CH_3)_3CBr + H_2O)$

- A) Ionization of alcohol to give carbocation
- B) Displacement of water from the protonated alcohol by bromide ion
- C) Protonation of alcohol
- D) Capture of a carbocation by bromide ion
- E) Loss of water from the protonated alcohol to give a carbocation

34-



- A) I
- B) ||
- C) III
- D) IV
- E) V

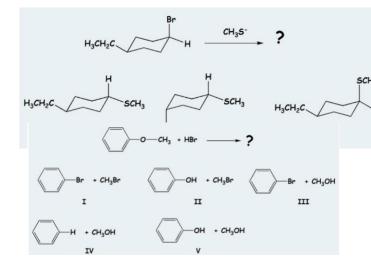
35-The product(s) of the following reaction is (are) :

- A) || & |||
- B) | & III
- C) II ONLY
- D) III ONLY
- E) I ONLY
- F)

36-The product(s) of the following reaction is (are):

- A) I
- B) II
- C) III
- D) IV
- E) V

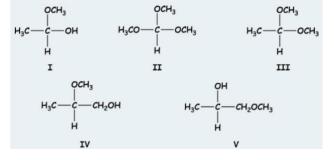




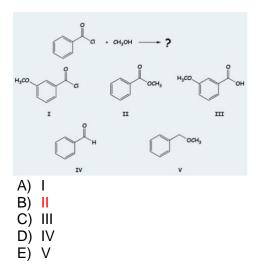
- B) II C) III D) IV
- E) V

38-Which of the following is hemiacetal :

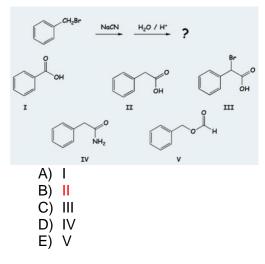
- A) |
- B) II
- Ć) III
- D) IV E) V











41-

